

## FURANOSESQUITERPENES FROM THE FRUITS OF *SMYRNIUM CORDIFOLIUM*

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**Key Word Index**—*Smyrnum cordifolium*; Umbelliferae; furanogermacranes; furanoeremophilane.

**Abstract**—In addition to two known furanogermacranes, a new furanogermacranes and a new and highly unstable furanoeremophilane were obtained from the fruits of *Smyrnum cordifolium*. The latter compound could be the precursor of a number of eremophilanolides obtained from some other *Smyrnum* species.

### INTRODUCTION

As a part of our continuing investigation of the genus *Smyrnum*, we have now isolated, in addition to known compounds, two new furanosesquiterpenes namely, 2- $\alpha$ -acetylfurodien and the highly unstable furanoeremophilane 1-oxo-10-hydroxyfuran-eremophila-3,7,11-trien from the fruits of *S. cordifolium* Boiss. The latter compound is probably the precursor of a number of eremophilanolides isolated from various *Smyrnum* species [1–4].

### RESULTS AND DISCUSSION

The known compounds glechomafuran (1) [5] and furodien (2) [6] were identified by comparing their spectra to those of known compounds. The new furanogermacranes, 2- $\alpha$ -acetylfurodien (3), was assigned the composition  $C_{17}H_{22}O_3$  on the basis of mass spectrometry ( $M^+$ ,  $m/z$  274, 63%). Its IR spectrum contained an acetyl band at  $1730\text{ cm}^{-1}$ , as well as furan bands at 1660, 1550 and  $880\text{ cm}^{-1}$ . Its  $^1\text{H}$  NMR spectrum confirmed the presence of an acetyl group ( $\delta$  2.08, 3H, s) and exhibited a furan proton signal at  $\delta$  7.06 (1H, s (br), H-12). Methyl signals were present at  $\delta$  1.40 (s, Me-15), 1.67 (s, Me-14) and 1.92 (d,  $J = 1\text{ Hz}$ , Me-13), whilst a three-fold doublet at  $\delta$  4.95 ( $J = 4, 11, 12\text{ Hz}$ ) showed the C-2 proton was geminal to the acetyl group indicating its equatorial character which was corroborated by the study of Dreiding models. Vinylic proton signals were at  $\delta$  5.22 (d (br), H-5) and 5.07 (d (br), H-1), and the doublets at  $\delta$  3.55 (1H, d,  $J = 16\text{ Hz}$ , H-9) and 3.42 (1H, d,  $J = 16\text{ Hz}$ , H-9') indicated the isolated methylene group. Other peaks were at  $\delta$  2.52 (1H, dd, H-3), 2.35 (1H, t, H-3'), 3.08 (1H, dd (br), H-6), 3.00 (1H, d (br), H-6'). There were four possible placements for the acetyl group in compound 3, namely C-2, C-3, C-6 and C-9. The three-fold doublet at  $\delta$  4.95 indicated that the acetyl group was situated at C-2 rather than at C-3, since in the latter case there would be a double doublet instead of three-fold doublet. On the other hand if the acetyl group was at C-9 there would have been a one proton singlet at around  $\delta$  5.00 and the isolated methylene group would not be seen. If it was at C-6 there would only be a doublet present around 5.00 ppm. In the MS the presence of the acetyl group and the furan moiety

was shown by peaks at  $m/z$  214 [ $M - 60$ ] $^+$  (base peak) and  $m/z$  108 (45%).

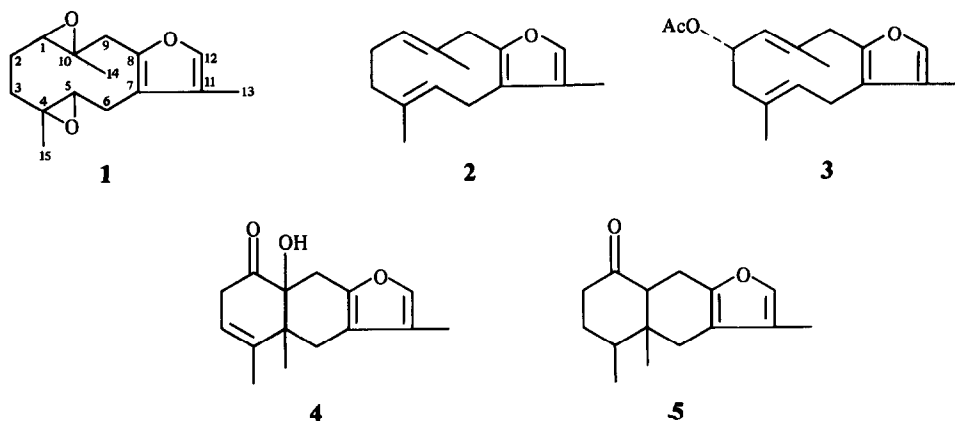
The new furanoeremophilane 4 had the composition  $C_{15}H_{18}O_3$  on the basis of mass spectrometry ( $M^+$ ,  $m/z$  246, 48%). Its IR spectrum showed a hydroxyl band at  $3450\text{ cm}^{-1}$  and a keto group at  $1725\text{ cm}^{-1}$  as well as furan bands at 1650 and  $870\text{ cm}^{-1}$ . The peak at  $\delta$  7.04 (1H, s, br) in the  $^1\text{H}$  NMR spectrum of 4 confirmed the presence of a furan ring. The chemical shifts of the methyl signals were present at  $\delta$  0.93 (s, Me-15), 1.71 (s, Me-14, vinylic methyl) and 1.98 (d,  $J = 1\text{ Hz}$ , Me-13, vinylic methyl) and indicated an eremophilane type compound. The vinylic proton signal at  $\delta$  5.55 (t,  $J = 8\text{ Hz}$ , H-3) was in agreement to that of H-3 in istanbulin E [3]. Since no conjugation was observed ( $\lambda_{\text{max}}$  218;  $\log \epsilon$  4.50) in its UV spectrum, the oxo group must be situated at C-1 as it is in istanbulins A–C, and E [1–3], other possible places for an oxo group were C-2, C-6 and C-9 which in each case would result in a conjugation. In the  $^1\text{H}$  NMR spectrum of 4 there was no peak between  $\delta$  3.5–5.5 corresponding to a hydrogen geminal to a hydroxyl group, therefore the latter must be tertiary and could only be situated at C-10; the location of a hydroxyl group at this position is quite common among *Ligularia* species [7–11].

Although a furanoeremophil-1-one (5) has been isolated from *S. olusatrum* by Bohlmann and Zdero [12] this is the first time we have isolated a furanoeremophilane from a *Smyrnum* species having structural resemblances to eremophilanolides obtained from other *Smyrnum* species. As all istanbulin type eremophilanolides have an oxygen function at C-1 (oxo group for istanbulins A and B [1], C [2] and E [3], hydroxyl for istanbulin D [3] and acetyl for istanbulin F [4]), and an ethyl group at C-4 (vinylic methyl for istanbulin E, exocyclic methylene for istanbulins C, D and F) the new compound could be the precursor of all these eremophilanolides.

### EXPERIMENTAL

The plant material was collected from southeastern Turkey (near Hakkari) in Sept. 1982 by M. Çoşkun (Faculty of Pharmacy, Ankara). A voucher specimen No. 10997 is deposited in the Herbarium of the Faculty of Pharmacy, University of Ankara.

*Isolation and characterisation of compounds 1–4.* Air dried and



powdered fruits of *S. cordifolium* Boiss. (100 g) were extracted with petrol (bp 40–70°)–Et<sub>2</sub>O (1:2). After filtration, the extract was concd *in vacuo* at room temp. and subjected to CC over silica gel (2 × 30 cm) with rapid elution. The fractions were collected within 2 hr, and the spectra of the compounds obtained from these fractions were recorded on the same day. Compounds 1 and 2 were obtained from the column as single compounds and crystallized from Et<sub>2</sub>O, whereas compounds 3 and 4 were separated on prep. TLC plates developed in petrol–Et<sub>2</sub>O (4:1). As observed in our earlier studies [13, 14] and by other workers [15], the use of CHCl<sub>3</sub> as a solvent causes a rapid oxidation of furanosesquiterpenes to their corresponding lactones, therefore <sup>1</sup>H NMR and <sup>13</sup>C NMR spectra were recorded soon after the compounds were dissolved in CDCl<sub>3</sub>.

1: Spectral data including X-ray data given in [16]. 2: Yield 120 mg, mp 45°, spectral data given in [6, 12], except <sup>13</sup>C NMR: (22.6 MHz, CDCl<sub>3</sub>): δ 128.9 (d) C-1, 39.5 (t) C-2, 40.9 (t) C-3, 121.9 (s) C-4, 127.8 (d) C-5, 24.3 (t) C-6, 119.0 (s) C-7, 149.9 (s) C-8, 27.3 (t) C-9, 134.5 (s) C-10, 121.9 (s) C-11, 136.1 (d) C-12, 8.9 (q) C-13, 16.2 (q) C-14, 16.5 (q) C-15. 3: Yield 15 mg, amorphous. UV λ<sub>max</sub><sup>MeOH</sup> nm: 218 (log ε 4.45); IR ν<sub>max</sub><sup>KBr</sup> cm<sup>-1</sup>: 3000, 2920, 2850, 1730, 1660, 1555, 1450, 1370, 1240, 1140, 1020, 880; <sup>1</sup>H NMR (NT-FT 200 MHz, CDCl<sub>3</sub>): text; MS (probe) 70 eV, *m/z* (rel. int.): 274 [M]<sup>+</sup> (63), 232 [M – Ac + H]<sup>+</sup> (30), 214 [M – AcOH]<sup>+</sup> (100), 108 [furan moiety]<sup>+</sup> (45). 4: Yield 18 mg, amorphous. UV λ<sub>max</sub><sup>MeOH</sup> nm: 218 (log ε 4.50); IR ν<sub>max</sub><sup>KBr</sup> cm<sup>-1</sup>: 3450, 2970, 2910, 2840, 1725, 1445, 1365, 1150, 1080, 1035, 880, 830; <sup>1</sup>H NMR (NT-FT 200 MHz, CDCl<sub>3</sub>): text; MS (probe) 70 eV, *m/z* (rel. int.): 246 [M]<sup>+</sup> (48), 228 [M – H<sub>2</sub>O]<sup>+</sup> (13), 205 [M – Me – CH=CH]<sup>+</sup> (100), 109 [furan moiety + H]<sup>+</sup> (78), 108 [furan]<sup>+</sup> (74).

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